

Contents lists available at ScienceDirect

Thermochimica Acta

journal homepage: www.elsevier.com/locate/tca



# Oxidative stability assessment of *[Bauhinia](http://www.elsevier.com/locate/tca) [purpurea](http://www.elsevier.com/locate/tca)* seed oil in comparison to two conventional vegetable oils by differential scanning calorimetry and Rancimat methods

Sarfraz Arain, S.T.H. Sherazi <sup>∗</sup>, M.I. Bhanger, Farah N. Talpur, S.A. Mahesar

*National Center of Excellence in Analytical Chemistry, University of Sindh, Jamshoro 76080, Pakistan*

# article info

*Article history:* Received 26 September 2008 Received in revised form 11 November 2008 Accepted 18 November 2008 Available online 27 November 2008

*Keywords:*

Differential scanning calorimetry Oxidative stability index *Bauhinia purpurea* seed oil Rice bran oil Cotton seed oil

# **ABSTRACT**

The differential scanning calorimetry (DSC) and oxidative stability index (OSI) techniques were applied for evaluating the oxidative stability of *Bauhinia purpurea*, rice bran and cotton seed oil. The DSC cell temperature was adjusted at four isothermal temperatures (110, 120, 130, and 140 ◦C) and oxidative DSC curves were obtained under oxygen flow rate of 50 ml/min. During the oxidation reaction the increase of heat was observed with sharp exothermic curve. *T*<sup>0</sup> was the oxidative induction time determined by the downward extrapolated DSC oxidative curve to time axis. The oxidation times (*T*0) acquired from each oil demonstrated a good correlation, *r* > 0.99, with the oxidative stability measured by Rancimat method. DSC method is concluded to be useful by consuming shorter time with lesser amount of sample.

© 2008 Elsevier B.V. All rights reserved.

## **1. Introduction**

The protection of oil quality, which remains suitable to consumers for longer time, is an important objective of quality control in the oil and fat industry. Shelf life of vegetable oils is the main characteristic that influence its suitability and market value [1]. The consequence of lipid oxidation leads to decrease in shelf life and has been recognized as the big problem in the food industry [2]. Oxidative stability is one of the most important indicat[ion](#page-2-0) [fo](#page-2-0)r maintaining the quality of the vegetable oils [3]. The resistance to oxidation is recognized as oxidative stability under d[iffer](#page-2-0)ent conditions and is expressed as the period of time necessary to accomplish an end point which can be selected according to diverse criteria, but typically leads to rapid increase in the rate of lipid oxidation is a measure of oxidative stability and [is](#page-2-0) [kn](#page-2-0)own as induction time [4,5]. A number of methods have been developed for the assessment of oxidative stability. There are various accelerated stability tests to quickly check the stability of oils and fats as oxidation is the major cause of oil degradation [6,7]. Usually, the Schaal oven test and the active oxygen (AOM) have been the most genera[lly](#page-2-0) [use](#page-2-0)d tests to evaluate oil stability [8]. It can also be determined by oxidative stability index (OSI) method as recommended by AOAC [9], which is widely used in the fat and oil industry by using two commercially available ins[trumen](#page-2-0)ts, the Rancimat from Metrohm Ltd. (Herisau, Switzerland) and the Oxidative Stability Instrument from Omniom Inc. (Rockland, MA) [10]. Recently, differential scanning calorimetry (DSC) has been used to determine the oxidative stability. Cross [11] was the first investigator who applied DSC, used isothermal conditions with flow of oxygen. The induction period was taken as the time where a fast exothermic reaction between oil and oxygen occurred[.](#page-2-0) [Seve](#page-2-0)ral researchers have used the application of thermal analysis for accelerated oil stability test [12,13]. Hassel's results showed that oil samples, which required 14 days via AOM, were evaluated in less than 4 h by DSC [14]. Kowalski with his coworkers and other researchers have also determined the oxidative stability of vegetable oils by DSC [15–18]. *Bauhinia purpurea* is a very valuable tree and has numerous tra[ditional](#page-2-0) [u](#page-2-0)ses. Naturally it is found in the sub-Himalayan tract and outer hills and valleys from the river Indus in Pakistan eas[twards](#page-2-0) to Assam and Myanmar, ascending to altitudes of 1500–1830 m [19]. In Pakistan the trees are cultivated in plains and su[b-mounta](#page-2-0)inous tracks, the fresh and dried flower buds of bauhinia are used as a food material. The leaves, stems and roots are widely used in Pakistan and in other countries for the treatment of several diseases especially infections, pain, diabetes, jaundice, leprosy, cough [and](#page-2-0) [a](#page-2-0)lso used in several medicine formulations [20]. Phytochemical and pharmacological studies of the plant revealed that the ethanol extract of leaves are used for analgesic, antipyretic, anti-inflammatory, antispasmodic and antimicrobial activity [21]. Several types of bioactive compounds were isolated from different parts of *B. purpurea* [22], and characterization [of](#page-2-0) [fatt](#page-2-0)y acids of Kachnar (*B. purpurea* L) seed oil has been done by Ramdan et al.[23]. The objective of the present work was to evaluate a[nd com](#page-2-0)pare the

<sup>∗</sup> Corresponding author. Tel.: +92 22771379; fax: +92 22771560. *E-mail address:* tufail.sherazi@yahoo.com (S.T.H. Sherazi).

<sup>0040-6031/\$ –</sup> see front matter © 2008 Elsevier B.V. All rights reserved. doi:10.1016/j.tca.2008.11.004

oxidative stability of *B. purpurea* with rice bran and cotton seed oil by DSC and OSI techniques.

### **2. Materials and methods**

*B. purpurea*, rice bran and cotton seed oil were used in the present study. The rice bran and cotton oil samples were obtained from local oil industry, Hyderabad, Sindh, Pakistan. Seeds from *B. purpurea* tree were collected from the campus of Sindh University, Jamshoro, Pakistan. The oil was extracted from seeds using hexane as a solvent through Soxhlet apparatus. All chemicals and reagents used were of the highest purity (Analytical grade) purchased from Merck (Darmstadt, Germany).

#### *2.1. Oxidative stability index*

Oxidative stability index was evaluated by OSI instrument (automated Metrohm Rancimat model 679) following AOCS Official Method Cd 12b-92 AOCS 1997 [9]. The instrument was run at 110 ◦C and an air flow rate of 201/h was bubbled through the oil  $(2.5 g)$ . The volatile degradation products were trapped in distilled water, increasing the water conductivity. The oxidative stability index was the time necessary [to re](#page-2-0)ach the conductivity curve inflection point.

# *2.2. Differential scanning calorimetry analysis*

The oxidative stability of conventional oils was determined by a Mettler Toledo differential scanning calorimeter DSC-820 (Schwerzenbach, Switzerland). The instrument was calibrated with pure indium and the baseline was obtained with an empty open aluminum crucible. The weighed amount of samples  $(5.0 \pm 0.25 \,\text{mg})$ were taken into open aluminum DSC crucible and placed in the sample compartment of the instrument. The four different isothermal temperatures were used (110, 120, 130, and 140 ◦C) and purified oxygen (99.99%) was passed through the sample at 50 ml/min.

# *2.3. Statistical analysis*

Statistical data were analyzed by using SAS 8.2 software (SAS institute, Cary, NC, USA). Duncan multiple range test to compare differences among means and SAS REG procedure was used between DSC  $T_0$  and OSI values. The relationship between DSC  $T_0$  and DSC isothermal temperature was also determined by the SAS REG procedure which is used for a simple linear equation. All the experiments were carried out in triplicate and reported as mean  $\pm$  standard deviation.

# **3. Results and discussion**

Straight line was observed with the stream of nitrogen (99.99%) flowing at 50 ml/min by the differential scanning calorimetry for *B. purpurea* oil at 130 °C as shown in Fig. 1, curve A clearly indicates that peak is not exothermic.While exothermic oxidation curve obtained when oil samples were run under oxygen atmosphere (99.99%) flowing at 50 ml/min (Fig. 1, curve B). Oxidation process is a principally exothermic reaction which occurs in between the oil and oxygen. The comparative stability of the *B. purpurea*, rice bran and cotton seed oil towards oxidation was analyzed by the extrapolated  $T_0$  values. Differential scanning calorimetry (DSC) oxidative induction time  $(T_0)$  and oxidative stability index (OSI) values of *B. purpurea*, rice bran and cotton seed oils are depicted in Table 1. OSI instrument at the isothermal temperature (110 $\degree$ C) gave significantly (*P* < 0.05) higher oxidative induction time than DSC technique. This variation may be due to a smaller sample size which was used in the DSC analysis as compared to OSI instru-



**Fig. 1.** A representative differential scanning calorimetry oxidation curve of *B. purpurea* oil: (A) isothermal curve at 130 ◦C with nitrogen (99.99%) flowing at 50 ml/min; and (B) isothermal curve at 130 ◦C with oxygen (99.99%) flowing at 50 ml/min.

**Table 1**

Differential scanning calorimetry (DSC) oxidative induction time (*T*0) and oxidative stability index (OSI) values of *B. purpurea*, rice bran and cotton seed oils.

Oil	DSC $T_0$ (min)				$OSI$ (min)
	110 $\degree$ C	$120^{\circ}$ C	$130^\circ C$	140 °C	110 $\degree$ C
B. purpurea Rice bran Cotton seed	483.33 132.89 172.41	269.77 72.74 92.00	99.11 36.06 42.74	48.44 18.66 20.30	1339.31 217.34 427.28

ment (5 mg vs. 5 g). The results of oxidative stability measured as induction time (IT) value by Rancimat assay for *B. purpurea* (Table 2) shows oxidative stability up to 1339.31 min, higher than rice bran (217.34 min) and cotton seed oil (427.28 min). The extraordinary stability of *B. purpurea* oil may be due to the presence of higher amount of tocopherols [24,25]. Each DSC isothermal temperature was found to have a significant effect (*P* < 0.01) on the DSC  $T_0$  (Table 1) measurements. For the analyzed oils, with increasing isothermal temperature a significant (*P* < 0.05) decrease was observed for  $T_0$ . Ge[nerally, w](#page-2-0)ith an increase in 10 $\degree$ C from 110 to 140  $\degree$ C, the  $T_0$  value was reduced approximately to half of its earlier appraisal (Table 1). This detail is in agreement with  $Q_{10}$  law for the association among the rate of chemical reaction and temperature [3,26]. An excellent coefficient correlation was found between the DSC  $T_0$  and OSI measurements as shown in Table 2. The coefficients of correlation were also highly significant (*P* < 0.01) for each evaluation. In observation of the high association between DSC *T*<sup>0</sup> and OSI time linear regression equations were calculated (Table 3). [T](#page-2-0)he regression equations of logarithm DSC  $T_0$  values against DSC isothermal temperature were established and given in Table 4. The coefficient of the determination  $(R^2)$  for analyzed oils was above 0.9956, showing good linear regression which revealed that oxidative stability of the oils can be accurately dete[rmined](#page-2-0) [b](#page-2-0)y DSC in a short time as compared to OSI method.

#### **Table 2**

Pearson correlation coefficient matrix between differential scanning calorimetry (DSC) and oxidative stability index (OSI) methods.<sup>a</sup>.



<sup>a</sup> Significance at 0.01 level (*P* < 0.01). DSC at isothermal temperature 110 °C, DSC120; DSC at isothermal temperature 120 ℃, DSC 130; DSC at isothermal temperature 130 ◦C, DSC140; DSC at isothermal temperature140 ◦C, OSI 110 ◦C.

#### <span id="page-2-0"></span>**Table 3**

Relationships between oxidative stability index (OSI) values and differential scanning calorimetry (DSC) oxidative induction time  $(T_0)$  at four different isothermal temperatures.



#### **Table 4**

Relationship between logarithm of DSC  $T_0$  values (log<sub>10</sub>  $T_0$ ) and DSC isothermal temperature (*T*) of *B. purprea*, rice bran and cotton seed oils.



### **4. Conclusion**

The results of present study for the determination of the oxidative stability of *B. purpurea* along with two conventional vegetable oils have shown a high correlation between DSC  $T_0$  values and OSI values. Both methods confirmed that *B. purpurea* oil is very stable oil when compared to rice bran and cotton seed oil. Due to considerable oxidative stability, *B. purpurea* oil may find some appropriate applications in future. Furthermore, the DSC method offers simplicity without using any chemical and time saving nature. Therefore, DSC method can be easily used as an alternative technique for the measurement of oxidative stability in edible oil processing industries where mostly OSI technique is used.

#### **References**

- [1] T.H. Smouse, Factors affecting oil quality and stability, in: K. Warner, N.A.M. Eskin (Eds.), Methods to Assess Quality and Stability of Oils and Fat-containing Foods, AOCS, Champaign, IL, 1995, pp. 17–36.
- [2] S.J. Jadhav, S.S. Nimbalkar, A.D. Kulkarni, D.L. Madhavi, Lipid oxidation in biological and food systems, in: D.L. Madhavi, S.S. Deshpande, D.K. Salunkhe (Eds.), Food Antioxidants Technological, Toxicological, and Health Prospectives, Marcel Dekker, New York, 1996, pp. 5–64.
- [3] C.P. Tan, Y.B. Che Man, J. Selamat, M.S.A. Yusoff, Food Chem. 76 (2002) 385–389.
- [4] J.P. Cosgrove, D.F. Church, W.A. Pryor, Lipids 22 (1987) 299–304.
- [5] E.A. Coppin, O.A. Pike, J. Am. Oil Chem. Soc. 78 (2001) 13–18.
- [6] P.J. White, Food Technol. 45 (1991) 75–80.
- [7] S. Paul, G.S. Mittal, Crit. Rev. Food Sci. Nutr. 37 (1997) 635–662.
- [8] P.J.Wan, Accelerated stability methods, in: K.Warner, N.A.M. Eskin (Eds.), Methods to Assess Quality and Stability of Oils and Fat-Containing Foods, American Oil Chemists' Society, Champaign, IL, 1995, pp. 179–189.
- [9] AOCS, Official Methods and Recommended Practices of the American Oil Chemists' Society, 5th ed., AOCS Press, Champaign, USA, 1997.
- [10] C.C. Akoh, J. Am. Oil Chem. Soc. 71 (1994) 211–216.
- [11] C.K. Cross, J. Am. Oil Chem. Soc. 47 (1970) 229–230.
- [12] C.P. Tan, Y.B. Che Man, Food Chem. 67 (1999) 177–184.
- [13] D.I. Cebula, K.W. Smith, J. Am. Oil Chem. Soc. 69 (1992) 298–992.
- [14] R.L. Hassel, J. Am. Oil Chem. Soc. 53 (1976) 179–181. [15] B. Kowalski, K. Ratusz, A. Miciula, K. Krygier, Thermochim. Acta 307 (1997) 117–121.
- [16] B. Kowalski, Thermochim. Acta 156 (1989) 347–358.
- [17] H. Gloria, J.M. Aguilera, J. Agric. Food Chem. 46 (1998) 1363–1368.
- [18] A. Raemy, I. Froelicher, J. Loelinger, Thermochim. Acta 114 (1987) 159–164.
- [19] M.S. Kaletha, B.P. Bhatt, N.P. Todaria, Allelopathy J. (1996) 247–250.
- [20] S.M. Morais, J.D.P. Dantas, A.R.A. Silva, E.F. Magalhaes, Rev. Bras. Farmacogn. (2005) 169–177.
- [21] K.L. Silva, M.W. Biavatti, S.N. Leite, R.A. Yunes, F. Monache, F.V. Cechinel, Z. Naturforsch. (2000) 478–480.
- [22] R.N. Yadava, P.A. Tripathi, Fitoterapia (2000) 88–90.
- [23] M.F. Ramadan, G. Sharanabasappa, Y.N. Seetharam, M. Seshagiri, J.T. Moersel, Food Chem. 98 (2006) 359–365.
- [24] H. Yoshida, J. Sci. Food Agric. 65 (1994) 331–336.
- [25] A. Kamal-Eldin, L. Appelqvist, Lipids 31 (1996) 671–701.
- [26] Mark Sewald & Jon Devries Food product shelf life, Medallian Laboratories, Analytical Progress, 2008, http://www.medlabs.com/file.aspx?FileID=91.